

## Some basic questions about ion channels (only a few addressed in rest of talk)

Concentration, flux, transit times vs. concentration outside channel, and applied external potentials. Both means and time variations are of interest.

Dependence of these quantities on interaction potentials, fixed charges in walls of channel, image charges caused by dielectric effects.

Explaining and modeling burst behavior.

Channel performance vs. protein structure.

Can the information transmission rate of ion channels be identifed?

Biologists can add dozens of additional questions.

Note: Engineers are fabricating ion channels (so far, larger diameters: 10 nm for engineers vs. 1 nm for biology)

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## Approaches to study

Measurements/experiments (measure conductance, identify protein structure through crystallography and modeling)

Molecular dynamics simulation: Track position of all the water molecules in addition to the ions. (Typically there are at least 100 water molecules per ion)

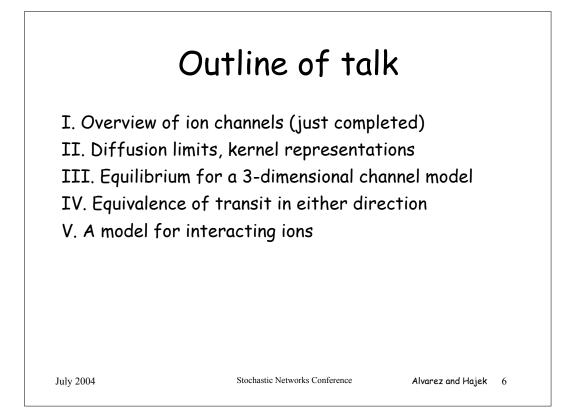
Brownian dynamics simulation: Track positions (and possibly velocities) of ions only. Interaction with water is accounted through diffusion coefficient (related to viscosity) and electric permittivity constant.

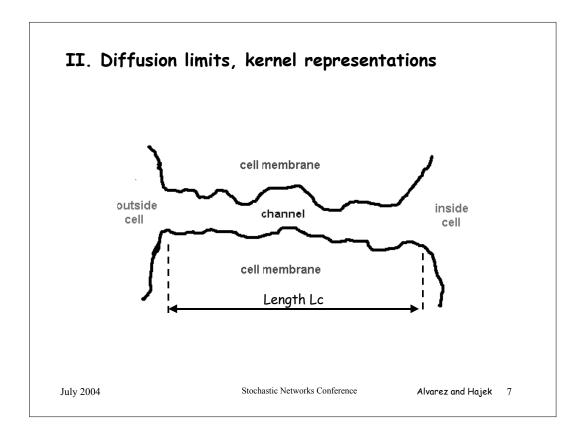
Partial differential equation approach: (Such as forward equations, but with incorporation of ion-ion interaction. Leads to Poisson-Nernst-Planck equations.)

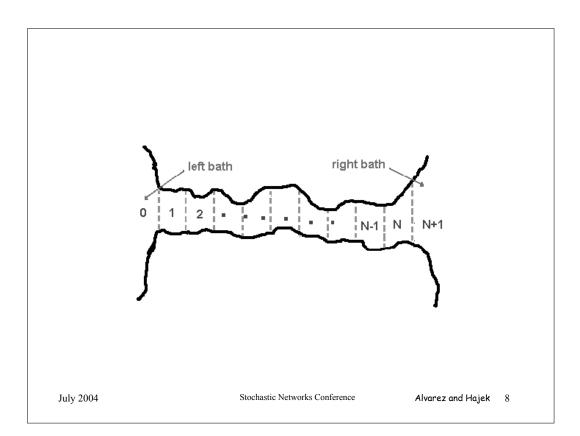
Reaction rate models (reduced complexity PDE approach)

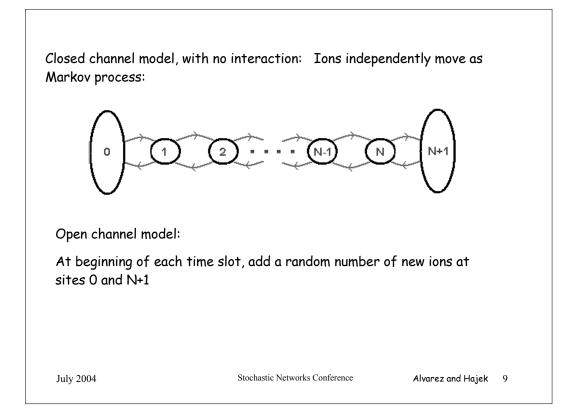
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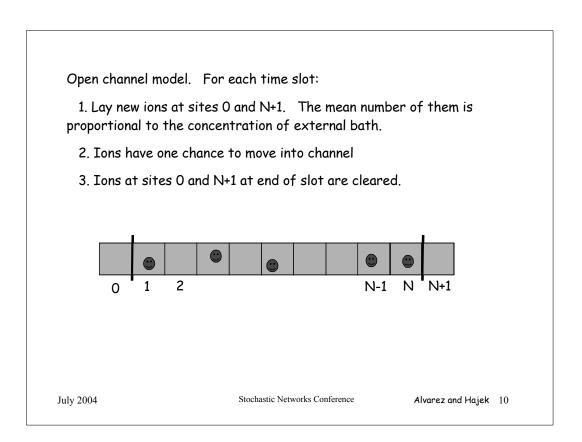
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## Relation between random walk and diffusion models for single ion (assuming no interaction among ions):

 $\mu(x), D(x)$  are drift and diffusion functions on  $[0, L_c]$ (Diffusion coefficient is often denoted  $\sigma^2(x)$  by probabilists, and D(x) by biologists, where  $\sigma^2(x) = 2D(x)$ .)

Random walk with jump probabilites:

$$\begin{aligned} r_i &= \quad \frac{\triangle t}{2\triangle x} \mu(\frac{iL_c}{N}) + \frac{\triangle t}{(\triangle x)^2} D(\frac{iL_c}{N}) \\ l_i &= \quad -\frac{\triangle t}{2\triangle x} \mu(\frac{iL_c}{N}) + \frac{\triangle t}{(\triangle x)^2} D(\frac{iL_c}{N}) \end{aligned}$$

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Alvarez and Hajek 11

**Diffusion limit**  $(\triangle x, \triangle t \rightarrow 0 \text{ with } \frac{(\triangle x)^2}{\triangle t} \text{ converging)}$  Limit diffusion can be described by a stochastic differential equation:

$$dx_t = \mu(x_t)dt + \sqrt{2D(x_t)}dw_t \quad \text{(Itô form)}$$
  
$$dx_t = \tilde{\mu}(x_t)dt + \sqrt{2D(x_t)} \circ dw_t \quad \text{(Stratonovich form)}$$

where  $\tilde{\mu} = \mu - (\nabla \cdot D)^T$  (in scaler case:  $\tilde{\mu} = \mu - \frac{\partial D}{\partial x}$ ) Forward equation:

$$\frac{\partial p}{\partial t} = \mathcal{L}^* p$$

where

$$\mathcal{L}^* p(x) = -\frac{\partial(\mu(x)p(x))}{\partial x} + \frac{\partial^2(D(x)p(x))}{\partial x^2} \quad \text{(Itô form)} \\ = \frac{\partial}{\partial x} \left\{ -\tilde{\mu}(x)p(x) + D(x)\frac{\partial p}{\partial x} \right\} \quad \text{(Stratonovich form)}$$

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Single Ion vs. System of Noninteracting Ions

The concentration at time t for a system of non-interacting ions is similar to the probability density at time t for a single ion.

In addition, an open system has ions entering from the boundaries. Leads, for example, to the following "kernel representation" of concentration at time t for the open channel model:

$$c(y,t) = \int_0^{L_c} c_0(x) p(t,x,y) dx + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_L(t-s) g(s,0,y) ds + D(0) \int_0^t c_R(t-s) g(s,0,y) ds + D(0) \int_0^t c_R(t-s) g(s,L_c,y) ds + D(0) \int_0^t c_R(t-s) g(s,0,y) ds + D(0) \int_0^t$$

where

• p(t, x, y) is the transition density of a single ion, killed at endpoints

• 
$$g(t, x, y) = \frac{\partial p(t, x, y)}{\partial x}$$

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Alvarez and Hajek 13

$$c(y,t) = \int_0^{L_c} c_0(x) p(t,x,y) dx + D(0) \int_0^t c_L(t-s) g(s,0,y) ds - D(L_c) \int_0^t c_R(t-s) g(s,L_c,y) ds$$

where

- p(t, x, y) is the transition density of a single ion, killed at endpoints
- $g(t, x, y) = \frac{\partial p(t, x, y)}{\partial x}$

Above describes the concentration for an open channel with noninteracting ions, operating in continuous time.

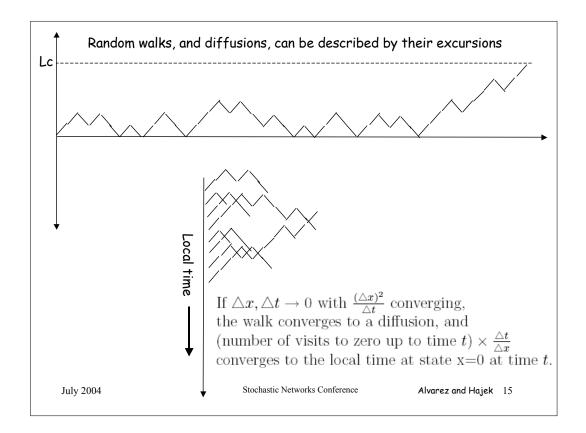
So certainly the concentration functions exist.

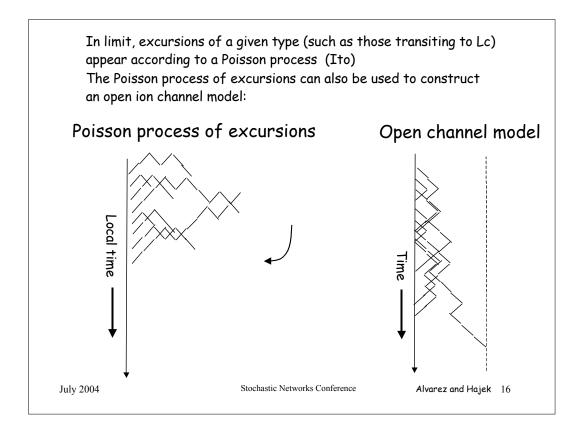
But does the channel model itself exist mathematically?

Yes -- see construction on next slide.

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## Details of construction of open channel model (Notation of Ikeda and Watanabe, Stochastic Differential Equations and Diffusion Processes, 1981)

- $\mathcal{U}$  denotes the set of continuous functions from  $R_+$  to  $[0, L_c]$  with initial value 0, endowed with the topology of uniform convergence on compact intervals.
- $\sigma(w) = \inf\{t : w(t) = 0 \text{ or } w(t) = L_c\} \text{ for } w \in \mathcal{U}.$
- $\mathcal{B}(\mathcal{U})$  is the Borel  $\sigma$ -algebra on  $\mathcal{U}$
- $\nu$  is the  $\sigma$ -finite measure on  $(\mathcal{U}, \mathcal{B}(\mathcal{U}))$  such that

$$\begin{array}{l}\nu(\{w:w(t_1)\in A_1,\ldots,w(t_k)\in A_k\})=\\ \int_{A_1}g(t_1,0,x_1)dx_1\int_{A_2}p(t_2-t_1,x_1,x_2)dx_2\cdots\int_{A_k}p(t_k-t_{k-1},x_{k-1},x_k)dx_k\end{array}$$

The definition of  $\nu$  is the same as the definition of  $n^+$  in Ikeda and Watanabe, except the function p is different.

- $N_p$  is a Poisson point process with time varying characteristic measure  $D(0)c_L(t)\nu$ . For example, for a time interval [a, b] and set  $U \in \mathcal{B}(\mathcal{U})$  with  $\nu(U) < \infty$ ,  $N_p([a, b] \times U)$  is a Poisson random variable with mean  $\int_a^b D(0)c_L(t)dt\nu(U)$ .
- $\beta_t = \int_0^t \int_{\mathcal{U}} I_{\{s \le t \le s + \sigma(w)\}} \delta_{w(t-s)} N_p(ds \, dw)$ . In words, the location of an ion arriving at time s with excursion w which is still in the channel at time t contributes a unit mass at location w(t-s).

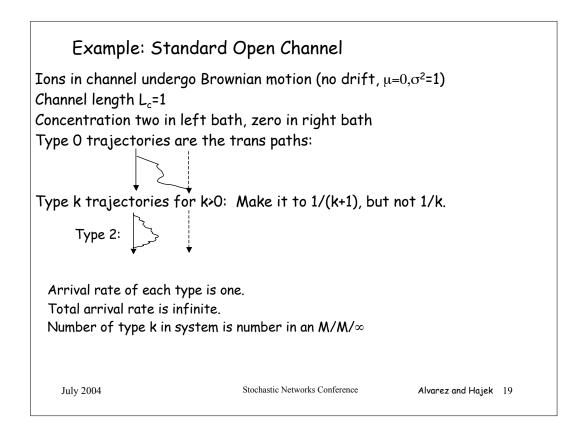
For such channel, the concentration is given by the expectation of the Poisson process. Thus,

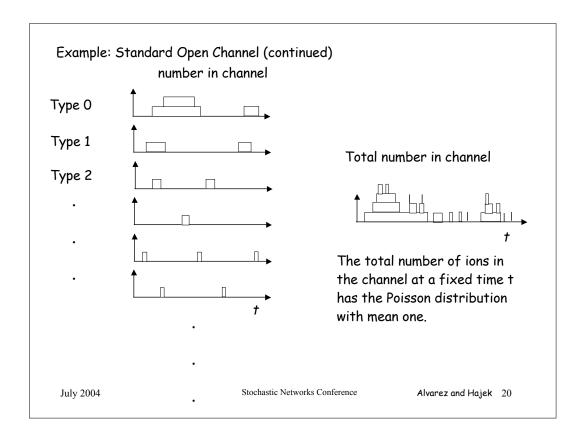
$$\begin{array}{lcl} c(y,t)dy &=& E[\beta_t(dy)] = E[N_p(\{(s,w) : s \le t \le s + \sigma(w), \ w(t-s) \in dy\})] \\ &=& D(0) \int_0^t c_L(t)g(t-s,0,y)dsdy \end{array}$$

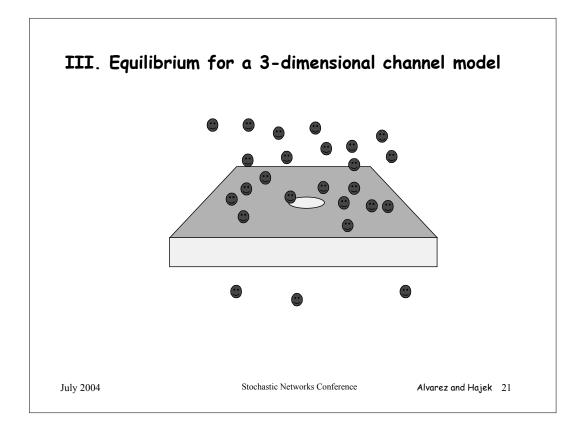
This explains in a direct way the terms arising from the boundaries in the kernel representation of the concentration density.

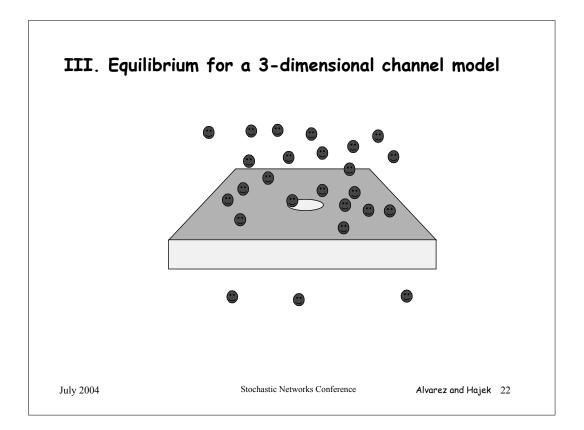
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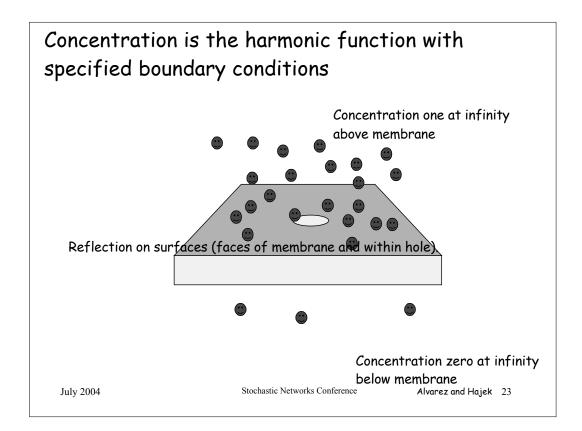
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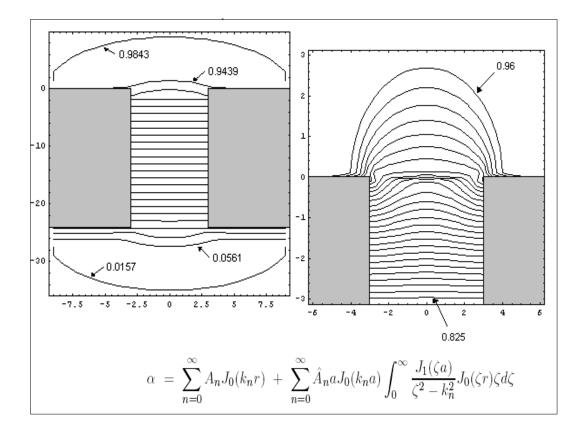












IV. Equivalence of transit in opposite directions No interaction model - Trans path equivalence				
"An interesting feature of the analytical mean passage times is that they are precisely symmetrical about zero voltage"				
* JAKOBSSON, E. AND CHIU, S. (1987) Stochastic theory of ion movement in channels with single-ion occupancy: Application to sodium permeation of gramicidin channels. <i>Biophys. J.</i> <b>52</b> , 33-45.				
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Explanation: For model with no interaction between ions: Time reversal maps the left-right trans-path distribution into the right-left trans path distribution.

Claim 1. For  $\gamma \in S_{LR}$ 

 $P_{LR}[\gamma] = P_{RL}[\gamma^r]$ 

Reversibility, or v symmetry (v\_iq\_{ij}=v\_jq\_{ji}) is key.

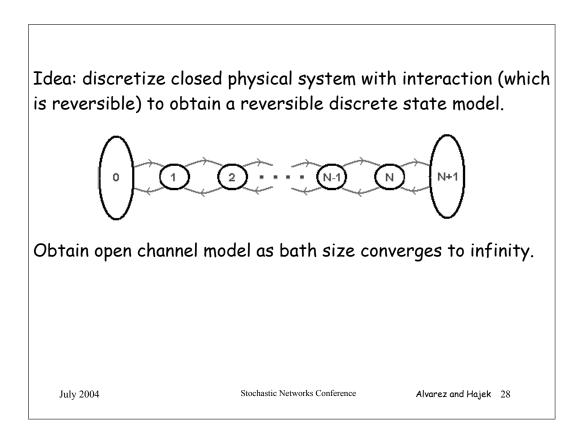
Implications?

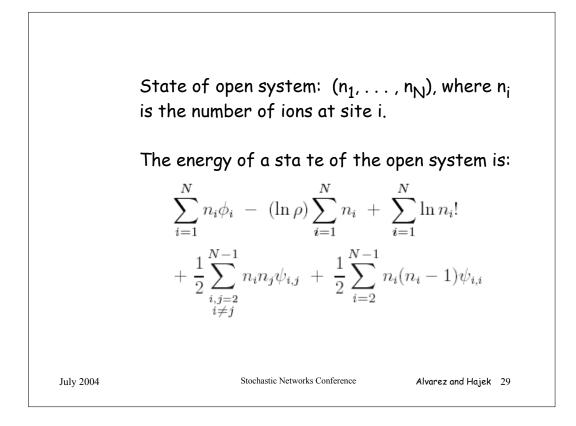
Is same true for models with ion-ion interaction?

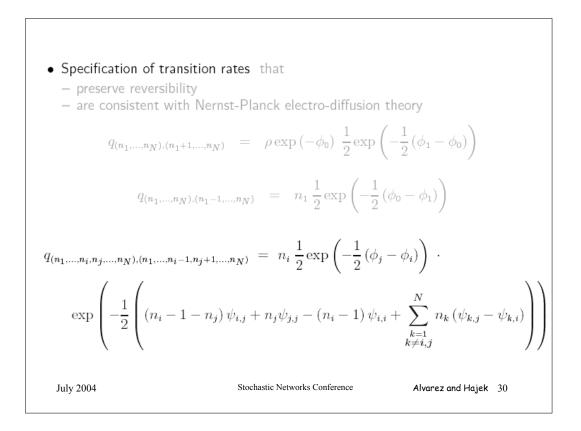
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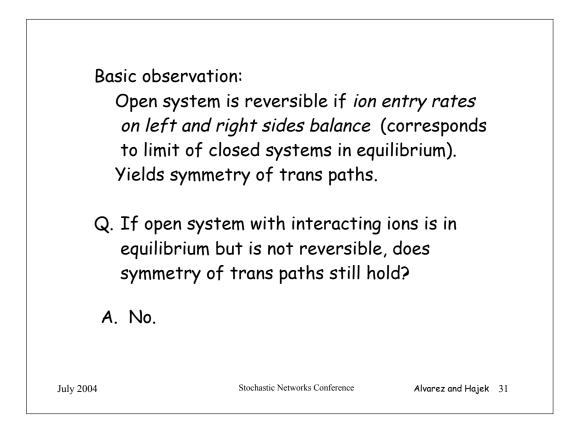
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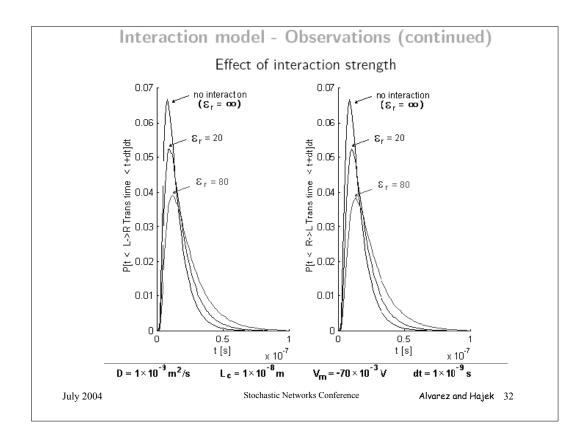
V. A model for	r interacting ions			
	f interacting ions subje action and viscosity is			
Therefore, expect symmetry of LR and RL trans path distribution to persist even when ions interact.				
Need model of motion.				
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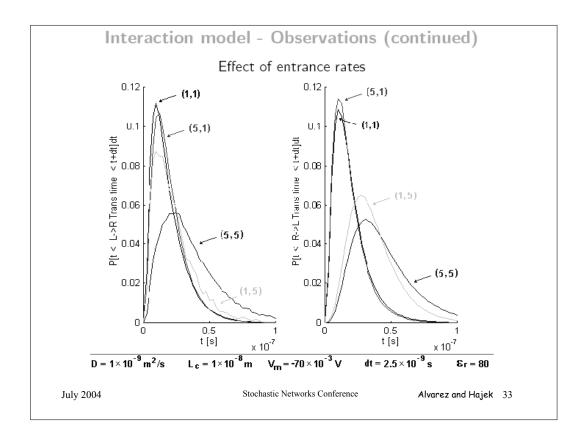


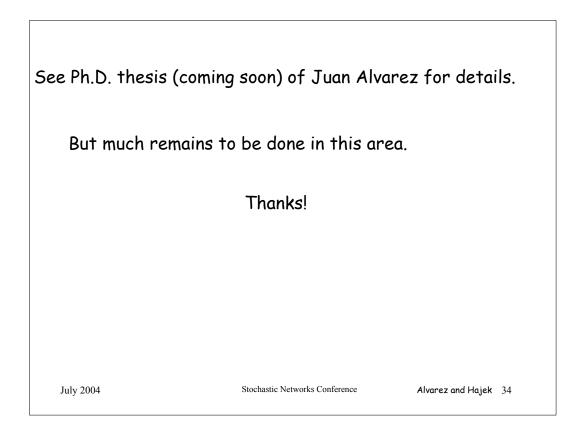












July 2004

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Alvarez and Hajek 35

July 2004

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